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CHAPTER STUDY GUIDE FOR CONTENT MASTERY Section 12.2 Stoichiometric Calculations In your textbook, read about mole-to-mole conversion. Read the following passage and then solve the problems. In the equation that follows each problem, write in the space provided the mole ratio that can be used to solve the problem.

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In your textbook, read about stoichiometry and the balanced equation. For each

statement below, write true or false. 1. The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry. 2. Stoichiometry is based on the law of conservation of mass. 3.

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Stoichiometry. The study of quantitative relationships between the amounts of reactants used and products formed by a chemical reaction; based on the law of conservation of mass. Actual Yield. The amount actually produced of a product. ... Chapter 5 Chemistry Study Guide. 28 terms.

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